

Modelling the hydrothermal liquefaction process

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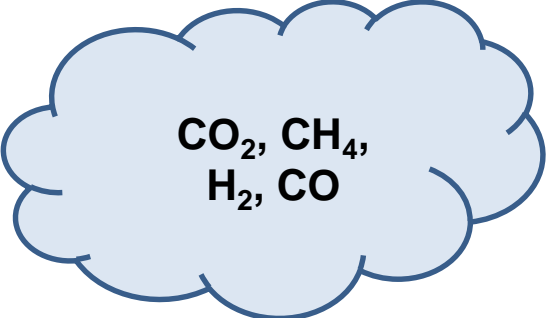
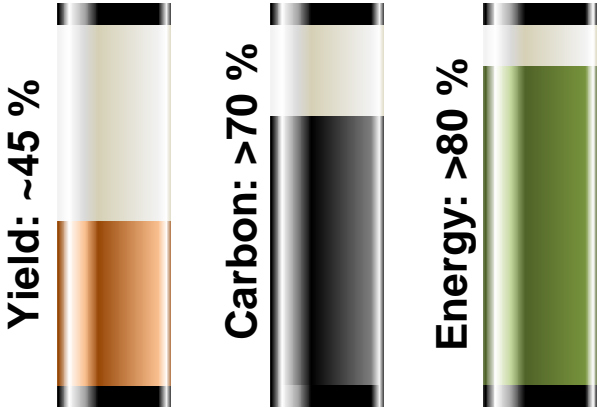
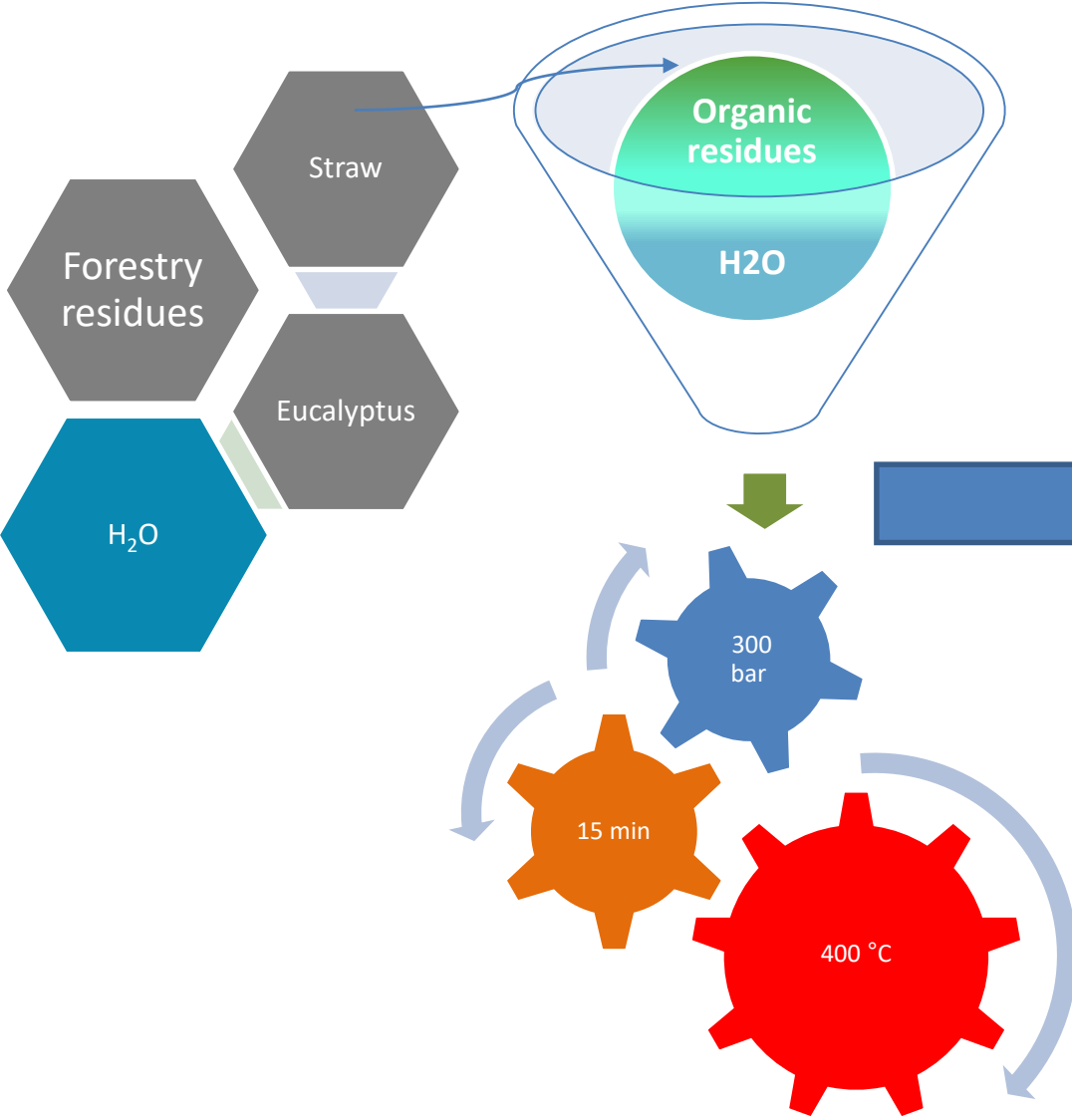
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What is HTL – the simple version



Gas production



Char & nutrients

Mimics the natural process for crude oil production.

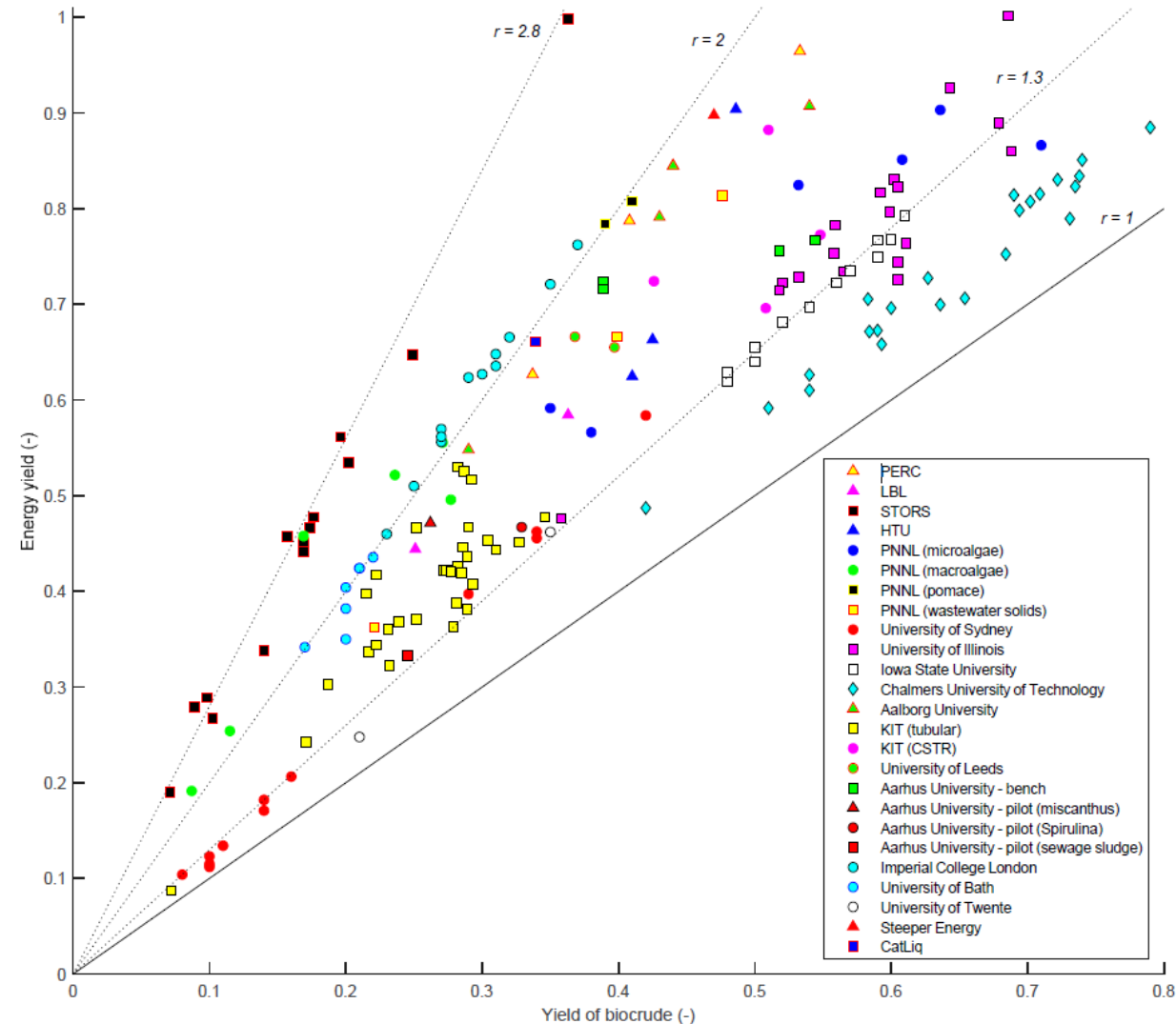
Time reduction: 100 mio. yrs vs. 15 min!

Biocrude ≠ crude oil

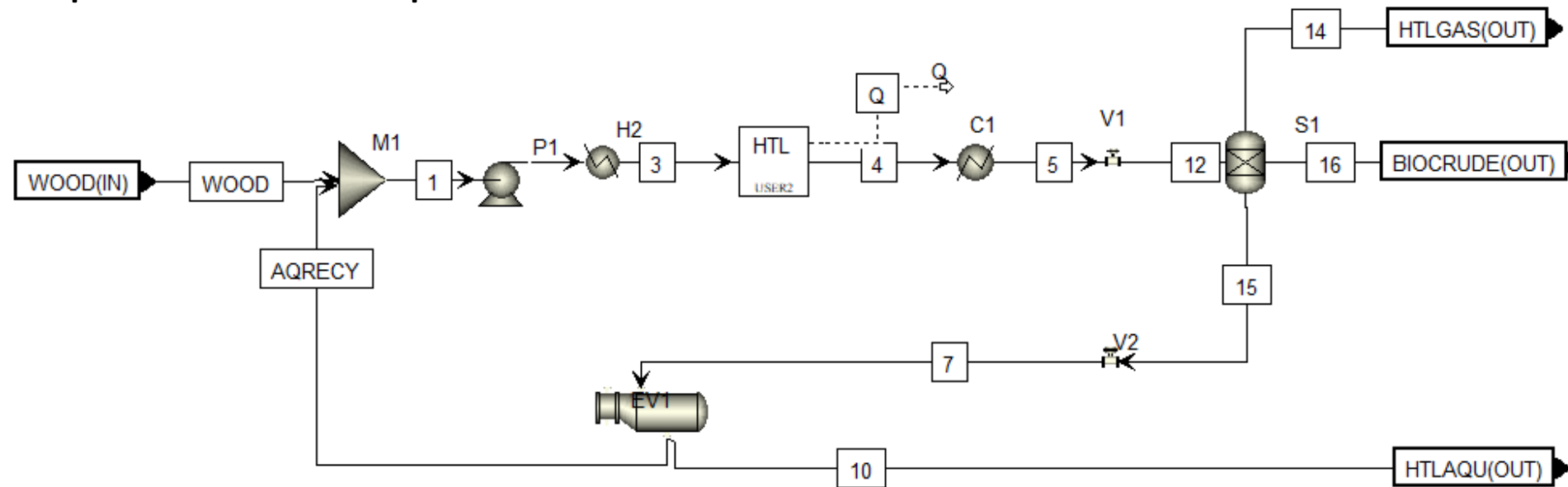


Biocrude yield and quality

- Biocrude composition dedicated by the feedstock.
- Energy ratio, $r = \frac{HHV_{bc}}{HHV_{feed}}$, follows iso-lines, determined by the feedstock
 - High HHV_{feed} ; low energy ratio
 - Low HHV_{feed} ; high energy ratio
- Yields can be greatly influenced by the process implementation



HTL process implementation

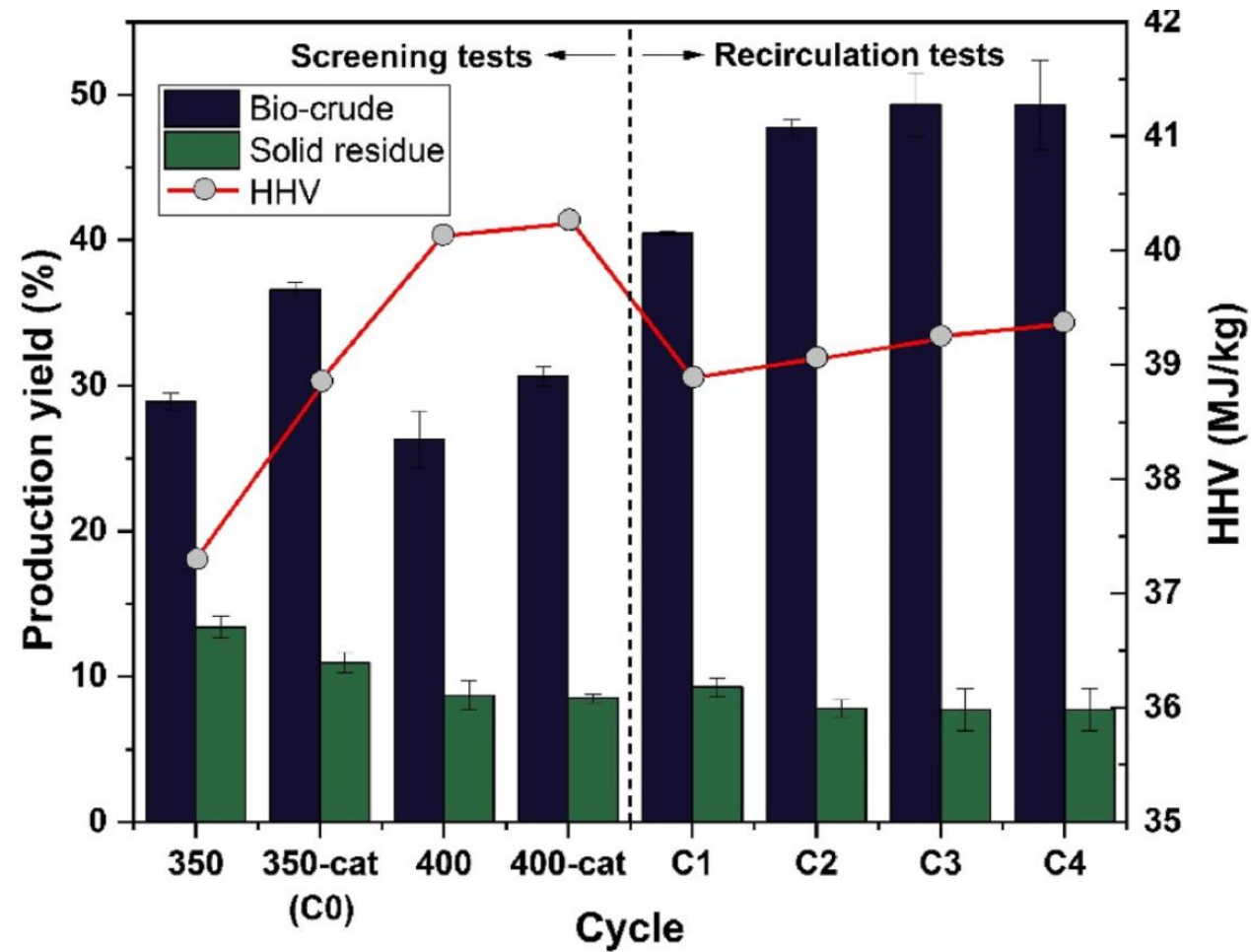


- Gas phase is combusted for internal heat recovery
- Bio-crude is demineralized and hydro-processed to fuels
- Aqueous organics can be recycled, waste water is processed by e.g. AD or HTG
- Important: Based on design data – HTL is not commercial!



Biocrude yield and quality

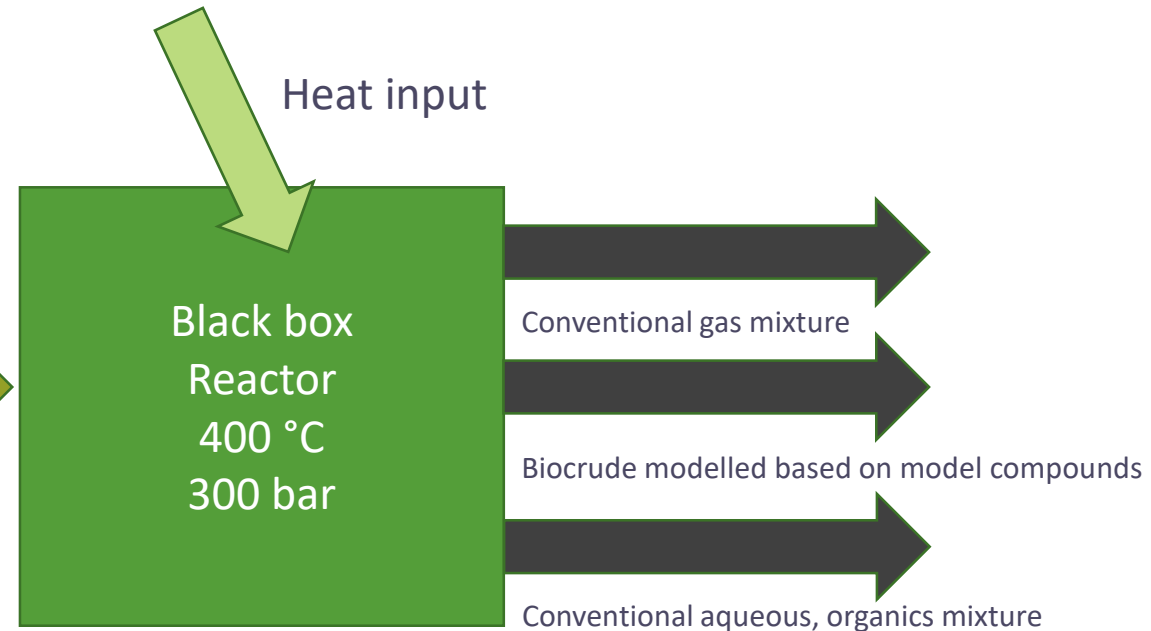
- Higher temperature, higher HHV
- Alkali agents (cat) increases carbon and mass yields
- Recirculating the aqueous phase organics increases biocrude yield.
- Energy recovery, $ER = Y \cdot \frac{HHV_{bc}}{HHV_{feed}}$, governed by the biocrude yield



HTL modelling in ASPEN+: General consideration in energy calculations

- The HTL chemistry of biomass is simply too complex to model!
- The pragmatic approach:

Non-conventional solid:
Ultimate & proximate analysis
Standard ASPEN+ model



- *Yield approach*, eventually coupled with predictive yield models calibrated by experimental data.



Modelling of biomass (in ASPEN+)

- Non-conventional solid (coal based, approx. 140 samples)
- Enthalpy and density models need to be specified:

$$h_{\text{wood}}^{\text{T,P}} = h_f^0 + \Delta h = h_f^0 + C_p(T) \cdot \Delta T$$

- Heat of formation
- Heat capacity
- *Heat of formation via the heat of combustion*

Option Code Number	Option Code Value†	Calculation Method	Parameter Names	Component Attributes
1 Heat of Combustion				
	1	Boie correlation	BOIEC	ULTANAL SULFANAL PROXANAL
	2	Dulong correlation	DLNGC	ULTANAL SULFANAL PROXANAL
	3	Grummel and Davis correlation	GMLDC	ULTANAL SULFANAL PROXANAL
	4	Mott and Spooner correlation	MTSPC	ULTANAL SULFANAL PROXANAL
	5	IGT correlation	CIGTC	ULTANAL PROXANAL
	6	User input value	HCOMB	ULTANAL PROXANAL
	7	Revised IGT correlation	CIGT2	ULTANAL PROXANAL

Option Code Number	Option Code Value†	Calculation Method	Parameter Names	Component Attributes
2 Standard Heat of Formation				
	1	Heat-of-combustion-based correlation	—	ULTANAL SULFANAL
	2	Direct correlation	HFC	ULTANAL SULFANAL PROXANAL COALMISC

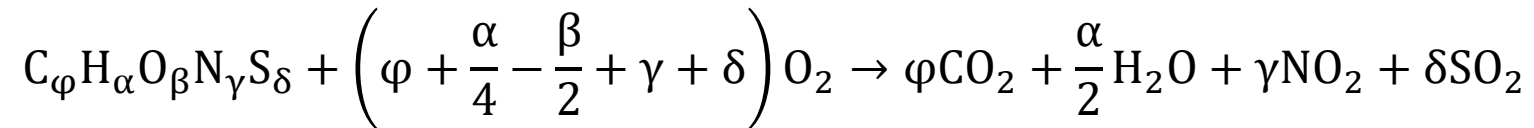
Option Code Number	Option Code Value†	Calculation Method	Parameter Names	Component Attributes
3 Heat Capacity				
	1	Kirov correlation	CP1C	PROXANAL
	2	Cubic temperature equation	CP2C	—



Modelling of biomass

- Calculating the enthalpy of formation, $\Delta h_{f,\text{biomass}}^{0,\text{daf}}$

Stoichiometric balancing the complete combustion reaction:



Consider the energy energy:

$$HHV = \Delta h_{f,\text{biomass}}^0 + \left(\varphi + \frac{\alpha}{4} - \frac{\beta}{2} + \gamma + \delta \right) \Delta_f h_{O_2}^0 - \varphi \Delta_f h_{CO_2}^0 - \frac{\alpha}{2} \Delta_f h_{H_2O}^0 - \delta \Delta_f h_{SO_2}^0 - \gamma \Delta_f h_{NO_2}^0 \left[\frac{J}{kg} \right]$$

Isolating and converting from molar to mass basis:

$$\Delta h_{f,\text{biomass}}^{0,\text{daf}} = HHV^{\text{daf}} - \left(3,278 * 10^5 w_C^{\text{daf}} + 1,418 * 10^6 w_H^{\text{daf}} + 9,264 * 10^4 w_S^{\text{daf}} - 2,418 * 10^4 w_N^{\text{daf}} \right) * 10^2 \left[\frac{J}{kg} \right]$$



Modelling of biomass

- Indirect methods in ASPEN+:

$$\Delta h_{f,\text{biomass}}^{0,\text{daf}} = \text{HHV}^{\text{daf}} - (3,278 * 10^5 w_{\text{C}}^{\text{daf}} + 1,418 * 10^6 w_{\text{H}}^{\text{daf}} + 9,264 * 10^4 w_{\text{S}}^{\text{daf}} - 2,418 * 10^4 w_{\text{N}}^{\text{daf}}) * 10^2 \left[\frac{\text{J}}{\text{kg}} \right]$$

- HHV is estimated from multiple empirical correlations, e.g. BOIE or DULONG formula

- Direct methods in ASPEN+:

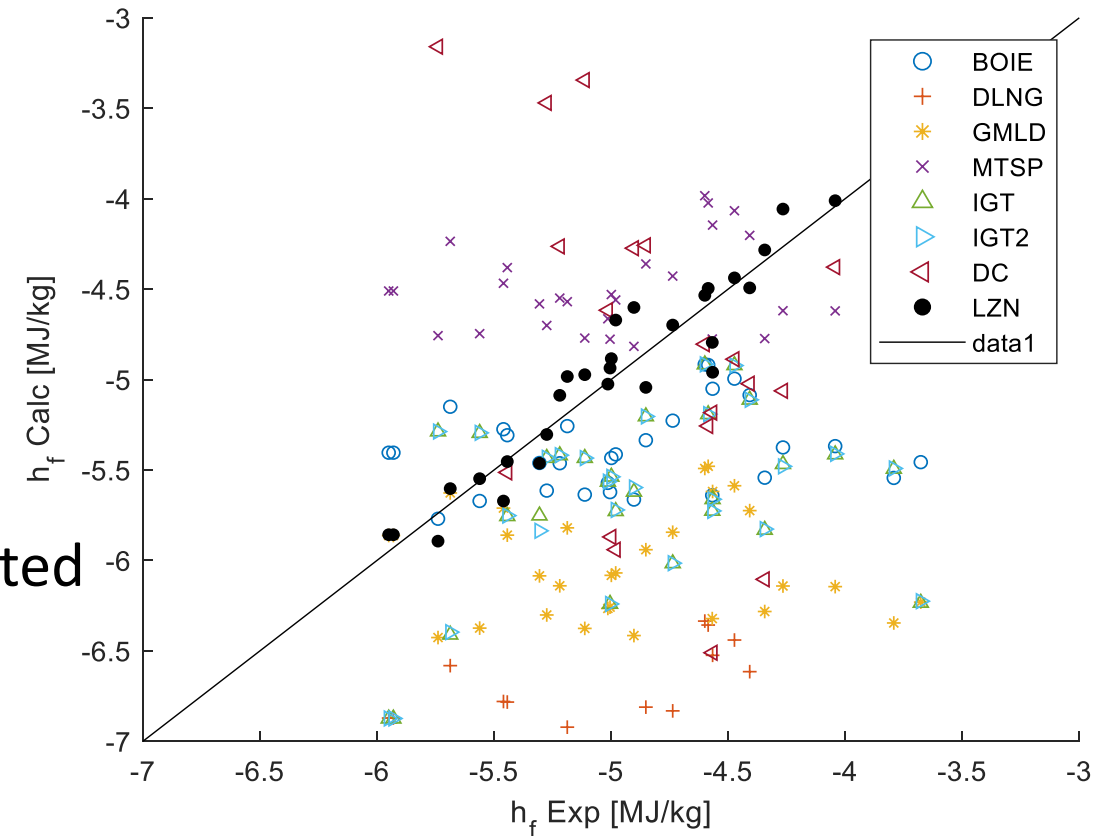
$$\Delta h_f^{0,\text{daf}} = \left[(a_1 w_{\text{C}}^{\text{dm}} + a_2 w_{\text{H}}^{\text{dm}} + a_3 w_{\text{H}}^{\text{d}}) 10^2 + (a_4 (w_{\text{C}}^{\text{d}} - w_{\text{FC}}^{\text{d}}) + a_5 w_{\text{VM}}^{\text{d}}) 10^2 \right]$$

- $\Delta h_{f,\text{biomass}}^0$ is estimated from proximate and ultimate analysis, solely, from an empirical correlation.

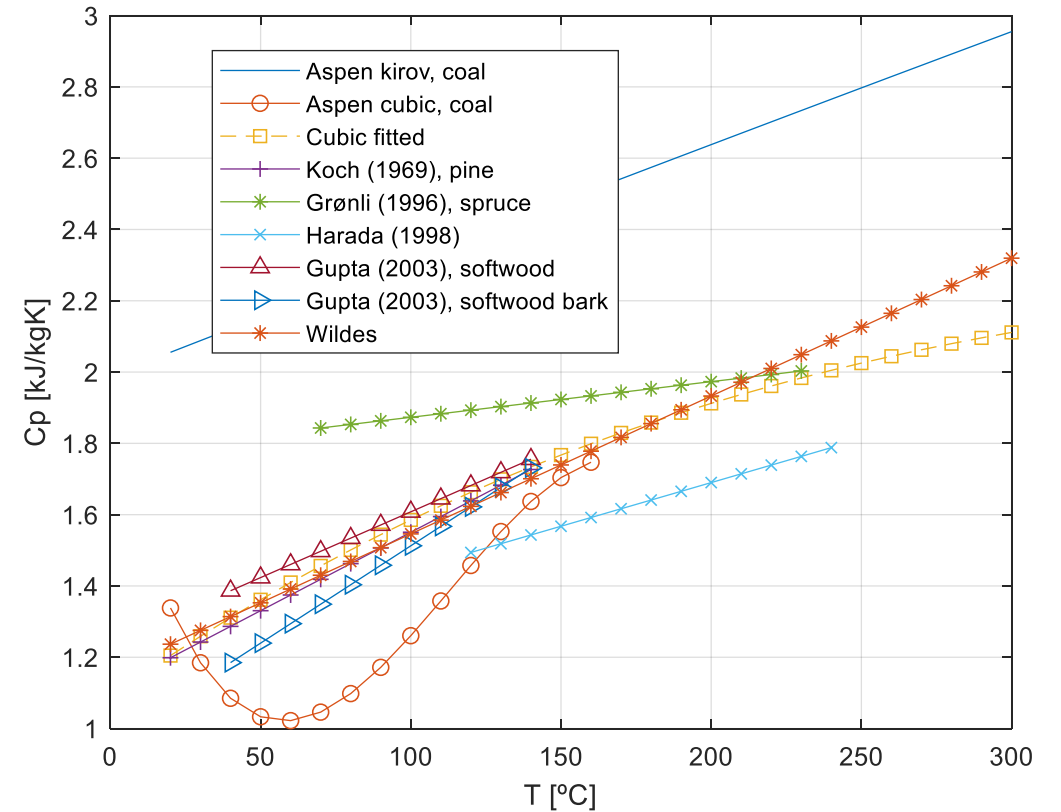
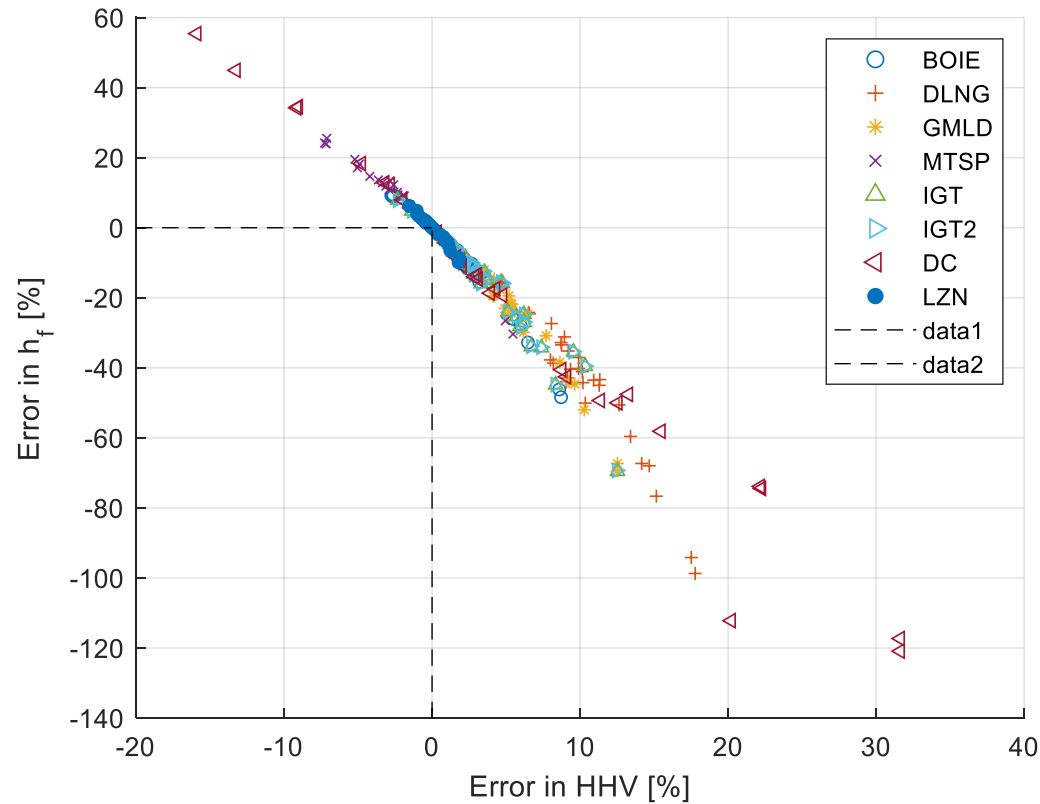


Modelling of biomass

- Verification using 20 different wood samples (Phyllis data base)
- All direct and indirect methods perform unsatisfactory.
- The solution: Fit your own *direct method!*
- Note: Correlation parameters can easily be imported to ASPEN+.



Modelling of biomass



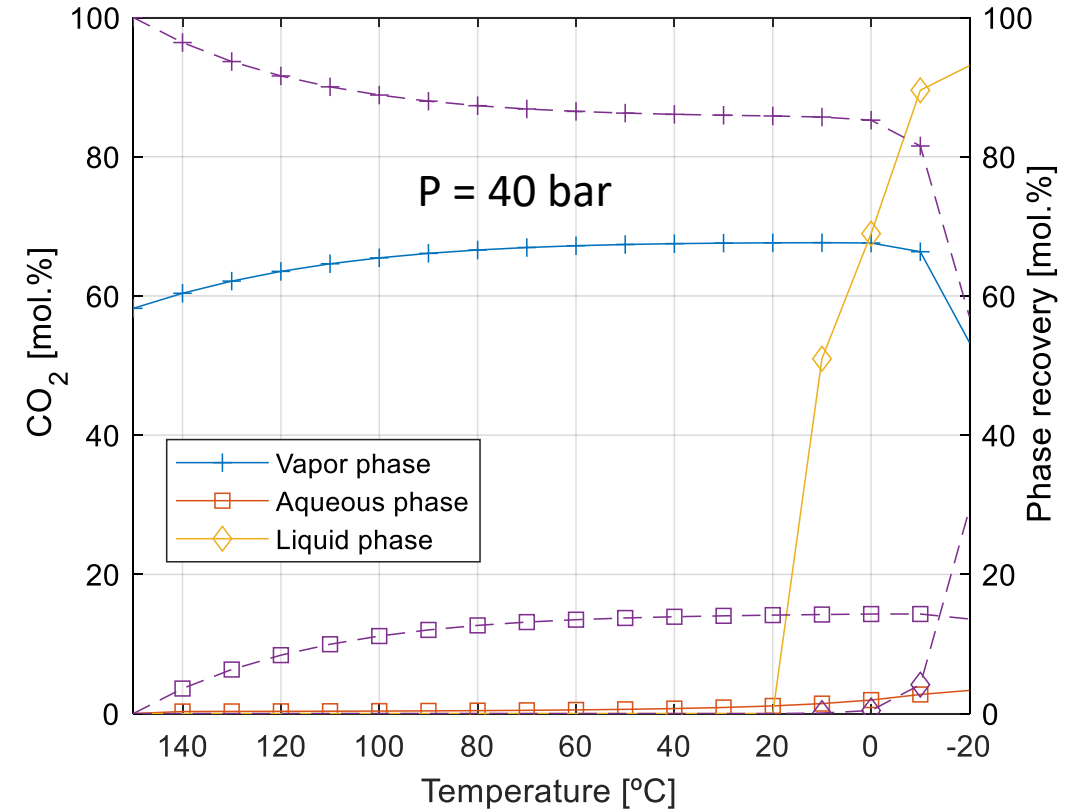
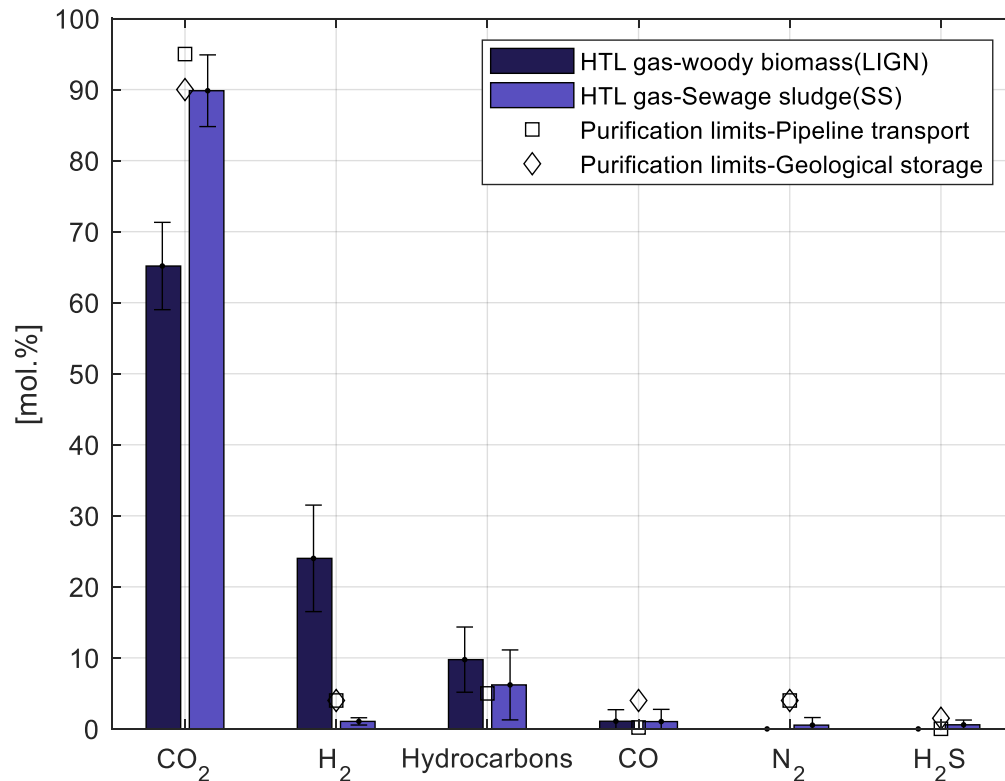
- Specific heat capacity of wood is NOT accurately calculated by the standard ASPEN+ correlation

$$h_{\text{wood}}^{T,P} = h_f^0 + \Delta h = h_f^0 + C_p(T) \cdot \Delta T$$



Gas phase considerations

- The gas phase is mainly CO₂. Modelled by a cubic equation or similar.
- Note; gas cooling at high pressure can initiate a VLE-situation.



Modelling of HTL biocrude

- Biocrude modelling is typically by a model compound approach
- "Finger printing" the biocrude by e.g. GC-MS analysis has limitations.

	mg/g of biocrude	Norm. Mass fraction
2-Cyclopenten-1-one, 2,3-dimethyl-	1.13	2.2%
Phenol	0.92	1.8%
2-Cyclopenten-1-one, 3,4,4-trimethyl-	1.27	2.5%
Caproic acid	0.50	1.0%
2-Cyclopenten-1-one, 3,4,4-trimethyl-	1.13	2.2%
2-Cyclopenten-1-one, 2,3,4,5-tetramethyl-	0.52	1.0%
2-Methylphenol	1.02	2.0%
Phenol, 4-methyl	2.29	4.5%
Heptanoic acid	0.91	1.8%
2,5-Dimethylphenol	1.07	2.1%
2,3-Dimethylphenol	2.76	5.4%
Phenol, 2,6-dimethyl	1.28	2.5%
Octanoic acid	1.09	2.1%
4-methyl catechol	7.20	14.1%
Benzeneacetic acid, 3-hydroxy	2.94	5.8%
Palmitic acid	4.92	9.6%
Myristic acid	2.38	4.7%
Octadecanoic acid	8.53	16.7%
SUM =	51.00	100%
% identified of the whole biocrude	5%	-

Modelling objectives

Bulk biocrude

Sum of model compounds

$$\text{HHV}_{\text{exp}} - \sum_{i=1}^n x_i \text{HHV}_i = 0$$

$$\text{C}(\text{wt}\%) - \sum_{i=1}^n x_i \text{C}_i(\text{wt}\%) = 0$$

$$\text{H}(\text{wt}\%) - \sum_{i=1}^n x_i \text{H}_i(\text{wt}\%) = 0$$

$$\text{O}(\text{wt}\%) - \sum_{i=1}^n x_i \text{O}_i(\text{wt}\%) = 0$$

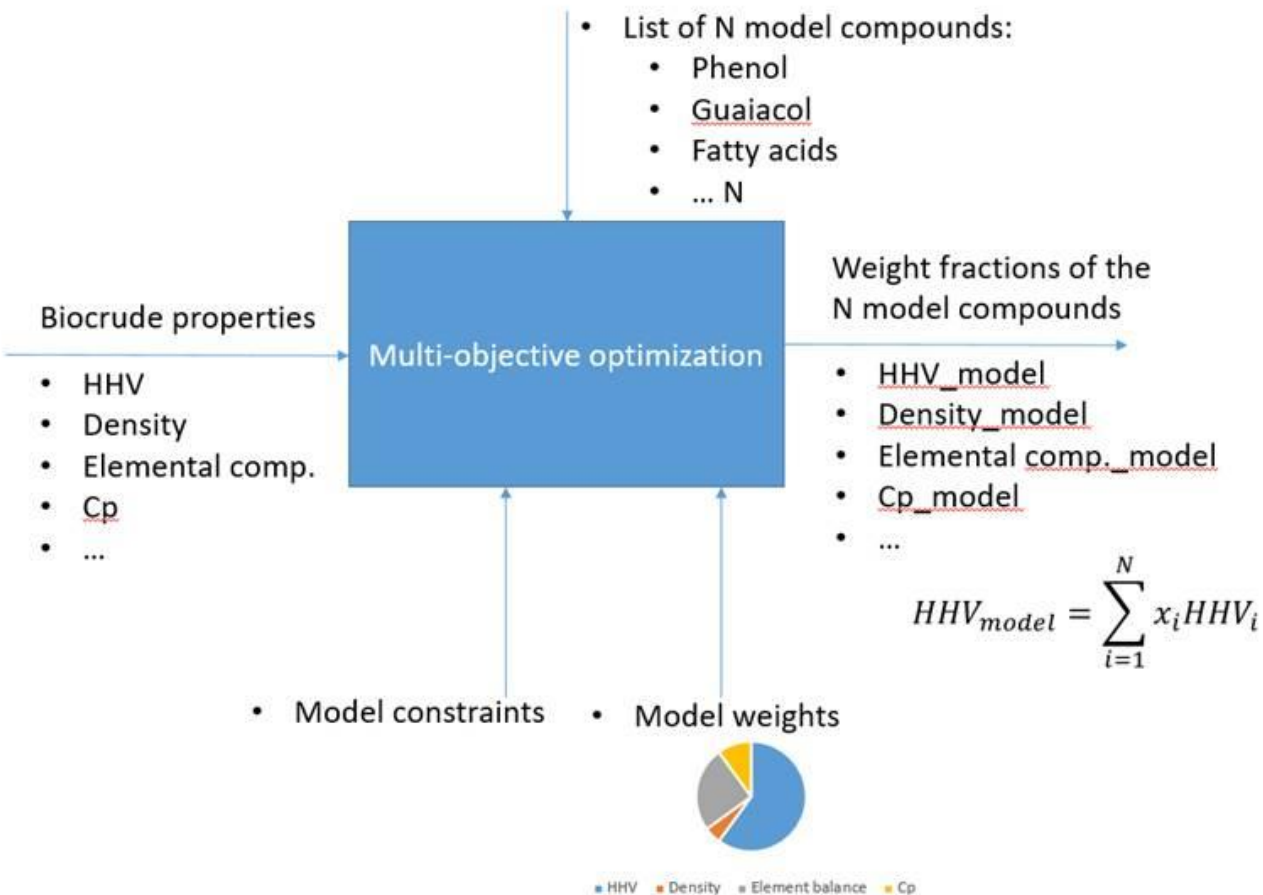
$$\rho_{\text{exp}} - \sum_{i=1}^n x_i \rho_{,i} = 0$$

$$\text{Cp}_{\text{exp}} - \sum_{i=1}^n x_i \text{C}_p = 0$$



Modelling of HTL biocrude

- Model compounds approach – selection of compounds



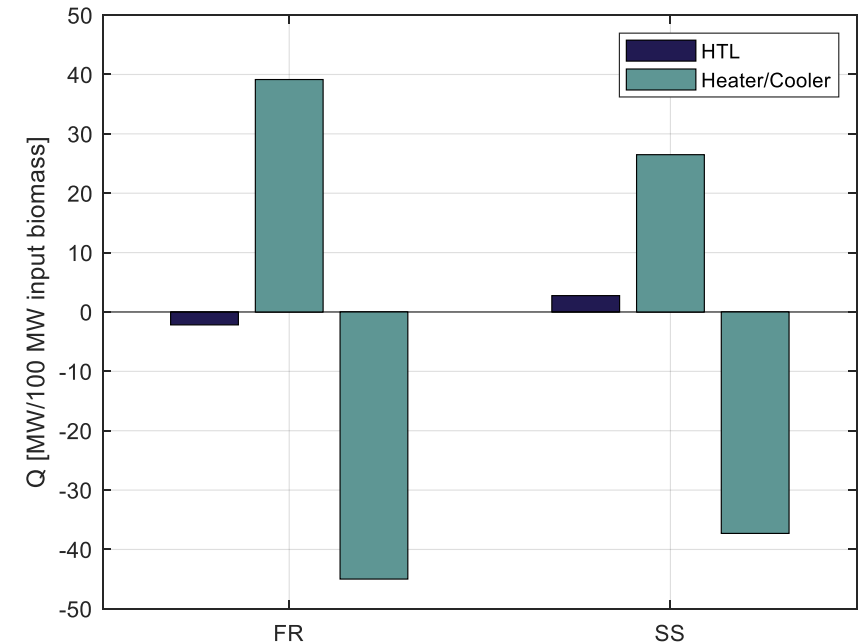
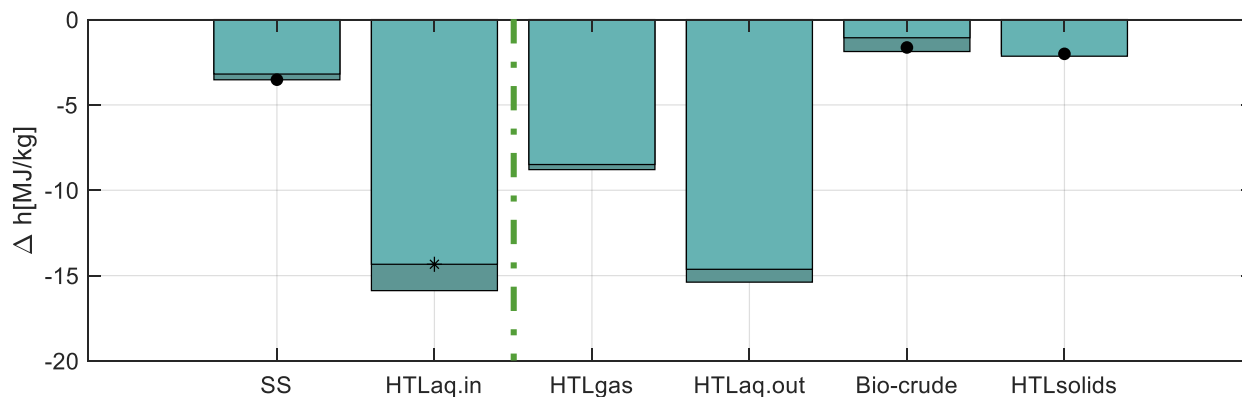
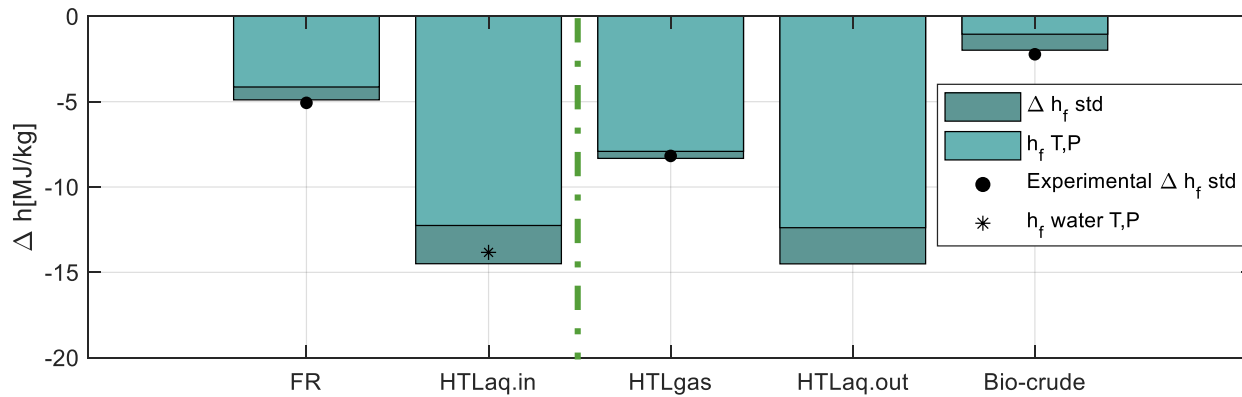
	Δh_f [MJ/kg]	HHV [MJ/kg]	C [wt.%] (daf)	H [wt.%] (daf)	O [wt.%] (daf)	Density [kg/m ³]
Reference	-2.22	35.90	80.00	8.40	11.00	1050.67
Initial values	-2.69	34.04	72.46	9.15	18.38	1007.19
Relative error [%]	20.99	5.18	9.42	8.95	67.12	4.14
Optimized (equal weights)	-2.24	37.75	77.16	10.36	12.48	1051.63
Relative error [%]	0.83	5.14	3.55	23.36	13.41	0.09
Optimized (Δh_f hard constraint)	-2.22	37.39	76.39	10.28	13.34	1047.83
Relative error [%]	0.00	4.15	4.52	22.37	21.23	0.27

The optimization routine is freely available from the authors:

Sanchez, E. M. L., Rosendahl, L., & Pedersen, T. H. (2019). Modeling of thermochemically liquefied biomass products and heat of formation for process energy assessment. *Applied Energy*



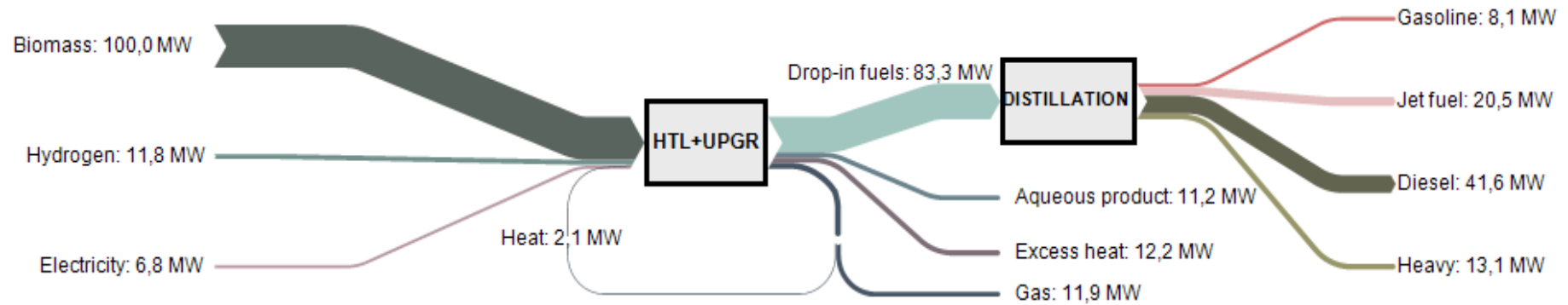
Some modelling results and considerations



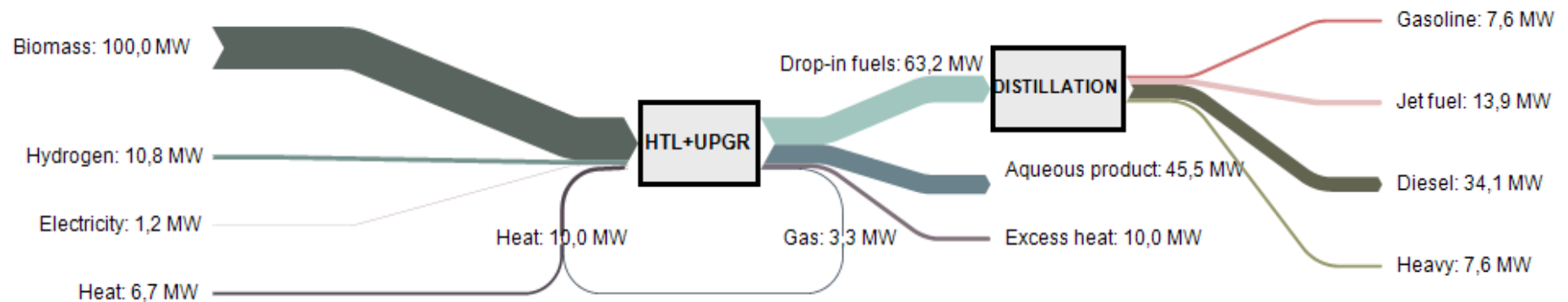
Case study: HTL integrated with refineries

- Biomass
- Hydrogen
- Electricity
- Heat
- Gas
- Drop-in fuels
- Aqueous product
- Excess heat
- Gasoline
- Jet fuel
- Diesel
- Heavy
- Methanol

Woody biomass

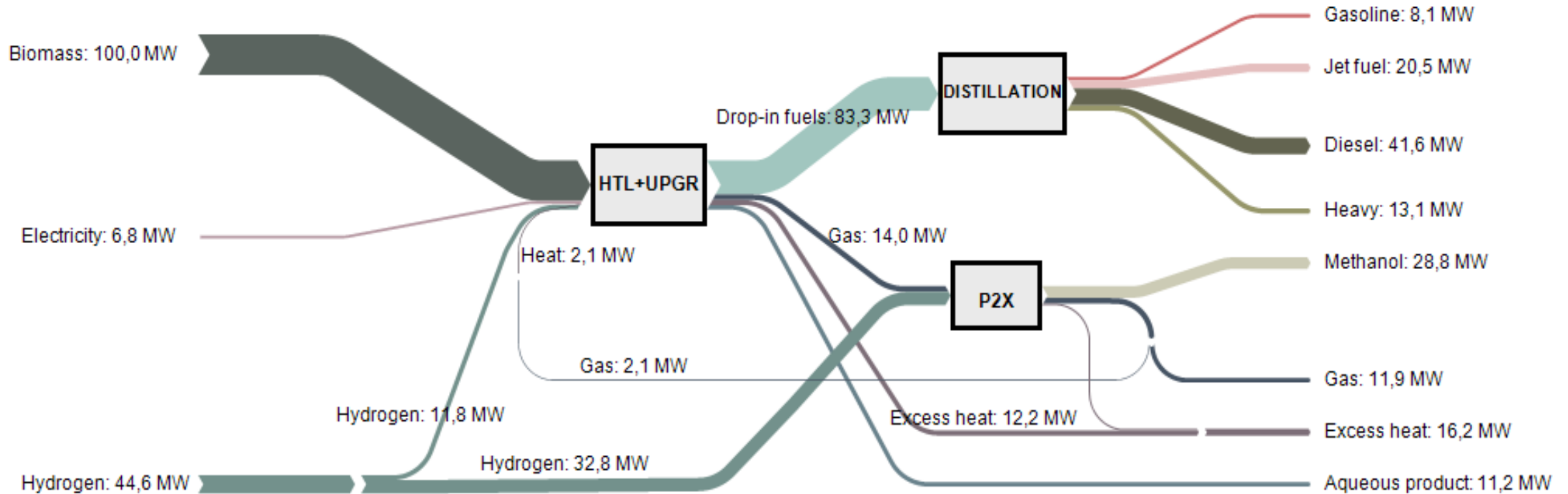


Sewage sludge



Case study: HTL integrated with refineries

HTL + eFuels (woody biomass)



Wrapping up

- The *non-conventional solid* approach is feasible for biomass modelling with fitted correlation parameters.
- Accurate enthalpy of formation calculations are of high importance.
- Cp correlations should be used with care. Always verify with experimental data.
- The *model compounds* approach with optimized bio-crude composition can accurately model bio-crude properties.
- Also be critical to your model results. 😊





Thank you for your attention!

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